रिक्त स्थानों की पूर्ति करें

a. CH_4 +..... $\xrightarrow{hv}$ $CH_3Cl + HCl$

b. $CH{=}CH_2 + HBr \longrightarrow$ .....

c. $CH_3 - CH = CH_2 + HBr \xrightarrow{Mark.}$ .....

d. $CH_3 - CH = CH_2 + HBr \xrightarrow{Per\ Oxide}$ .....

e. CH_3CH_2OH +..... $\xrightarrow{anh.\ ZnCl_2}$ ..... + H_2O

f. $CH_3CH_2OH + PCl_3 \longrightarrow$ .....+ H_3PO_4

g. $CH_3CH_2OH + PCl_5 \longrightarrow$ .....+ HCl + ...

h. $CH_3CH_2OH + PBr_5 \longrightarrow$ .....

i. $CH_3CH_2CH_2OH + SOCl_2 \xrightarrow{.....}$ ..... + ... + SO_2

j. $CH_3COOAg + Br_2 \xrightarrow{.....} CH_3Br + AgBr$ + ...

k. $C_6H_5N_2Cl \xrightarrow{CuCl/HCl}$ ..... + ... + ...

l. $C_6H_5N_2Cl + HBF_4 \longrightarrow$ ..... + ... + ...

m. $CH_3CH_2Cl + 2\ Na \xrightarrow{dry\ ether}$ .....